



## 2019 Lower Long Lake Water Quality Review

### Introduction

The goals of this testing protocol were to monitor various water quality parameters of the lake, compare results to historical data, and identify any potential risks to the health of Lower Long Lake. Water samples were taken at four different locations and tested for 14 parameters. Tests were conducted on a monthly basis from April through August. Tests were conducted with a Hanna Multiparameter Water Quality Meter or LaMotte SMART2 Colorimeter.

Test results were compared to historical data from the report “2018 Lower Long Lake Water Quality Review” by LakePro, Inc.

In this report, we included historical data from Water Quality Investigators. Their report provided annual averages for many of the parameters from 2002 to 2009. Including this data allows us to see more accurate trends in the water quality data. In order to make the analysis easier, we added annual averages for our data and trendlines on the graphs. The trend lines allow us to see the direction each water quality parameter is moving.

### Results

<u>Parameter</u>	<u>2019 Season Average</u>	<u>Target Range</u>	<u>Status</u>
Temperature	69.5 °F	Less Than 75 °F	● Healthy
Dissolved Oxygen	7.7 mg/L	4.0 – 12.0 mg/L	● Healthy
Total Phosphorus	104 ppb	0 – 100 ppb	● Slightly High
Phosphate	52 ppb	0 – 100 ppb	● Healthy
Nitrate	359 ppb	0 – 1,000 ppb	● Healthy
Chlorophyll-a	4.8 ppb	0 – 7.3 ppb	● Healthy
Transparency	18.5 feet	More than 6.5 feet	● Healthy
pH	7.7 S.U.	7.0 – 9.0 S.U.	● Healthy
Total Dissolved Solids	378 ppm	0 – 1,000 ppm	● Healthy
Conductivity	757 ppm	0 – 1,500 ppm	● Healthy
Alkalinity	129 ppm	100 – 250 ppm	● Healthy
Sulfate	11.1 ppm	3 – 30 ppm	● Healthy
Fluoride	0.09 ppm	0.01 – 0.30 ppm	● Healthy
Chloride	138 ppm	0 – 230 ppm	● Healthy





Parameter	July 2019 Result	Target Range	Status
Cadmium	Non-Detect	0 – 0.005 ppm	● Healthy
Chromium	Non-Detect	0 – 0.1 ppm	● Healthy
Copper	0.3 ppm	0 – 1.3 ppm	● Healthy
Lead	Non-Detect	0 – 0.015 ppm	● Healthy
Nickel	Non-Detect	0 – 0.1 ppm	● Healthy
Zinc	Non-Detect	0 – 5.0 ppm	● Healthy
Mercury	Non-Detect	0 – 0.002 ppm	● Healthy
Cyanide	Non-Detect	0 – 0.2 ppm	● Healthy
Arsenic	Non-Detect	0 – 0.01 ppm	● Healthy

## Preface

2019 was LakePro's ninth year testing water quality on Lower Long Lake. The historical data reveals trends over the testing history. The trend lines on the following graphs show the change from 2002 to 2019. Each successive year of testing will provide more insight into how the lake changed.

Each test represents a snapshot of the water quality when the sample was pulled. Water quality parameters can change from morning to night, day to day, or year to year. The discussion below will focus on the results listed above. We drew conclusions from the data, timing, and weather, but it is important to understand that each successive year of testing will help support trends and averages and improve our discussion.

This year's weather was marked by heavy rainfall in May and June, followed by much drier conditions in July and August. The precipitation can have two effects: dilute and flush what is in the lake and deliver new substances with runoff. The drier conditions along with the heat of late summer tend to promote plant growth and concentrate what is already in the lake. The discussion below describes how these processes may have affected the water quality results.

## Discussion

The results of this year's testing indicate that the water of Lower Long Lake remained very healthy throughout the summer. The results show that the aquatic environment was very suitable to support natural wildlife. Also, the lake was safe for recreational uses, such as swimming, boating, fishing, etc., as there are no signs of pollution.

The **Temperature** of the surface water was cool during the rains, then increased sharply in July. Colder water can hold more oxygen, so the higher temperatures created concern about dissolved oxygen in the lake, especially during the heat of summer. Despite the higher temperatures, the **Dissolved Oxygen** in the lake remained healthy throughout the summer. Furthermore, the sufficient oxygen concentration eliminated any concerns headed into winter when ice seals the lake off from atmospheric oxygen.

The concentrations of **Total Phosphorus** were fluctuated in and out of the target range throughout the summer. **Phosphate**, the usable form of phosphorus, was in the target range during all testing events. Lakes naturally accumulate more nutrients over time, so these slightly higher concentrations may be indications the nutrients are pooling in the lake.





9353 Hill Road • Swartz Creek, MI 48473  
(810) 635-4400 • Fax (810) 635-4404

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The **Nitrate** concentrations remained in the lower half of the target range throughout the entire summer. Although concentrations are still in the target range, it is important that residents take measures to ensure their property is not contributing excess fertilizers to the lake.

We tested water samples for **Chlorophyll** as a direct indicator of plant production. The chlorophyll concentrations responded strongly to the water temperatures, increasing sharply in July.

The **Transparency** was excellent during every test, measuring deeper than any other lake we tested last year. Transparency can be affected by many different factors, including suspended solids, dissolved solids, acids, and algae growth. The clear water is generally a positive attribute, but it also allows more sunlight to reach the lake bottom to fuel plant growth.

The **pH** was within the target during all testing events.

The **Total Dissolved Solids** and **Conductivity** were within their target ranges and decreased from spring to summer. This usually occurs in most lakes as precipitation slows, delivering fewer external substances to the lake, allowing more to be filtered and flushed out of the lake.

The **Alkalinity** was within the target range all summer and seemed to fluctuate with the rainfall. The major reason for the decrease in the late summer was productivity. As lake organisms become more active in the summer time, they produce more carbon dioxide. As this gas dissolves, it needs to be buffered, using up the carbonate ions.

The **Sulfate, Fluoride, and Chloride** were all within their target ranges for the entire summer.

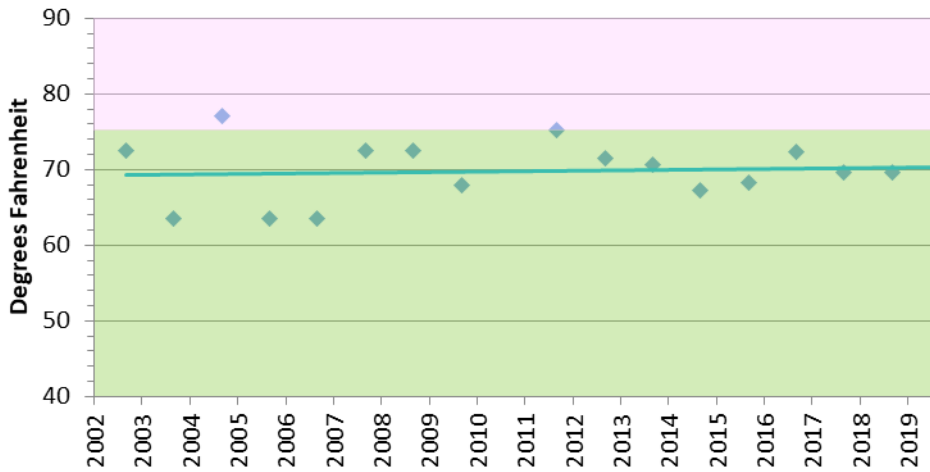
The late summer test for **Heavy Metals** was consistent with previous years. There was a latent concentration of copper ions in the water, but all other parameters were non-detectable in the water.





## Historical Trends

### Lake Temperature

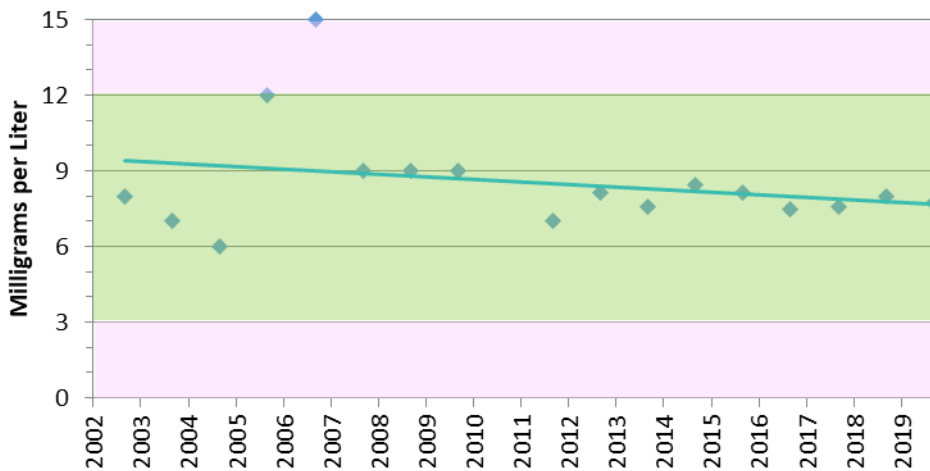


Temperature (°F)					
	April	May	June	July	August
Site 1	49.1	66.1	70.8	80.5	78.7
Site 2	49.9	66.7	71.3	80.5	79.0
Site 3	50.3	67.1	71.8	80.3	79.0
Site 4	51.0	67.2	71.4	80.2	79.2
Season Average					69.5

#### Target Range: < 75°F

The historical trend for temperature is generally flat. The temperature was affected by the dates selected for testing and the weather each year. As we collect data in subsequent years, the trend line should become a more accurate indicator of the changes in the lake.

### Dissolved Oxygen

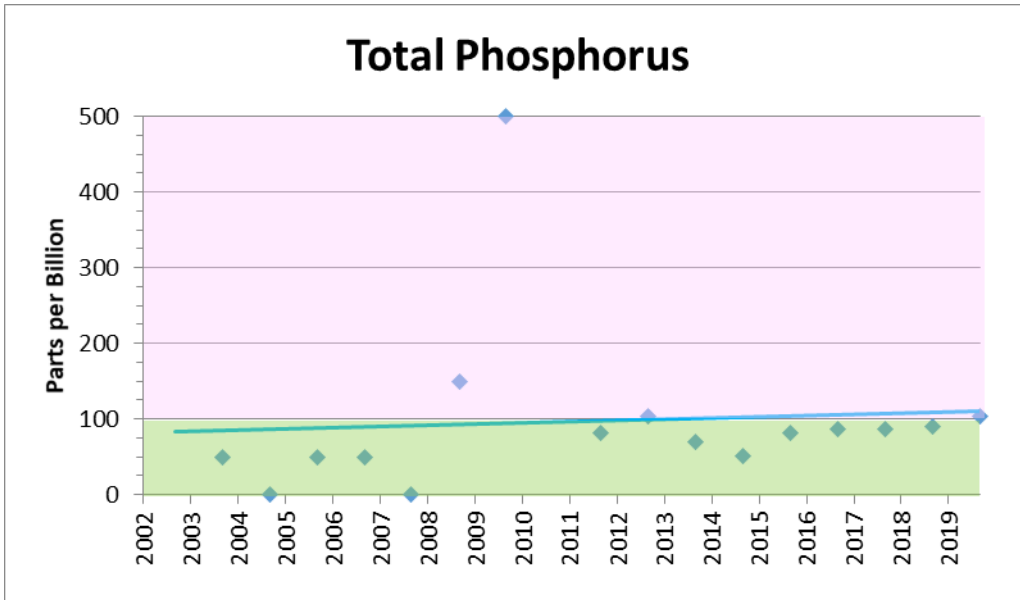


Dissolved Oxygen (mg/L)					
	April	May	June	July	August
Site 1	9.9	7.9	7.0	6.7	6.3
Site 2	9.6	8.1	7.3	7.2	6.1
Site 3	9.8	8.4	7.6	6.8	6.1
Site 4	9.9	8.2	7.7	7.1	6.4
Season Average					7.7

#### Target Range: 4.0 – 12.0 mg/L

Despite steady temperatures, the dissolved oxygen trend showed a slight decrease over the testing history. Oxygen concentrations remained very healthy, showing the lake carries a healthy oxygen concentration despite changes in temperature. We will continue to watch this trend and recommend further actions if the decrease steepens.

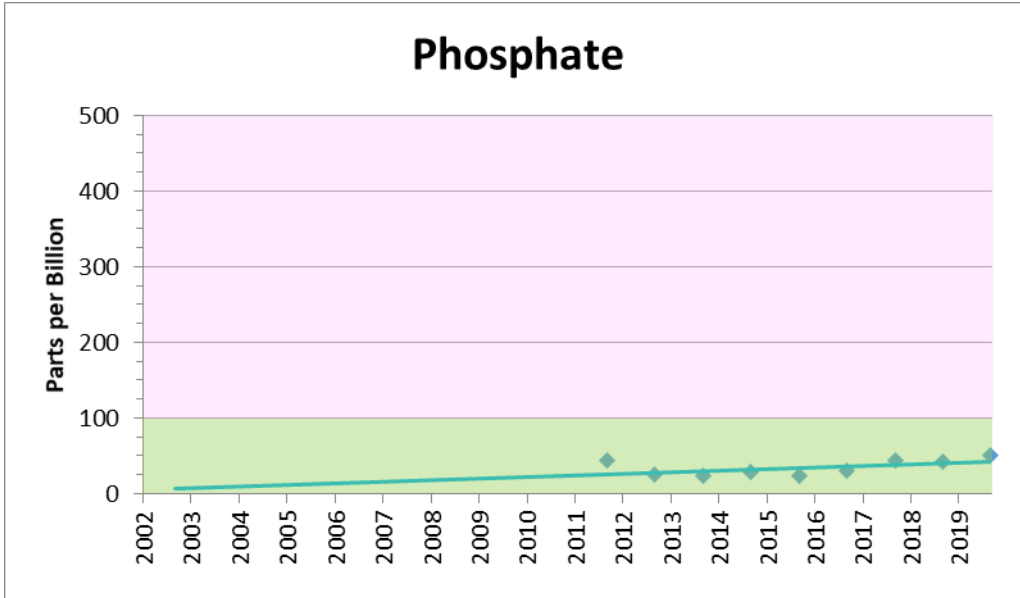




Total Phosphorus (ppb)					
	April	May	June	July	August
Site 1	120	90	110	120	120
Site 2	90	100	100	100	100
Site 3	80	120	110	110	90
Site 4	110	90	110	110	90
Season Average					104

**Target Range: 0 – 100 ppb**

The total phosphorus annual averages showed a slight increase over the testing history, due primarily to high results in 2008 through 2010. Lakes generally accumulate substances, including nutrients, in the process of eutrophication. The state law banning phosphorus fertilizers and active plant management will help decrease the phosphorus in the future. The phosphorus concentrations were exemplary for a heavily developed watershed.

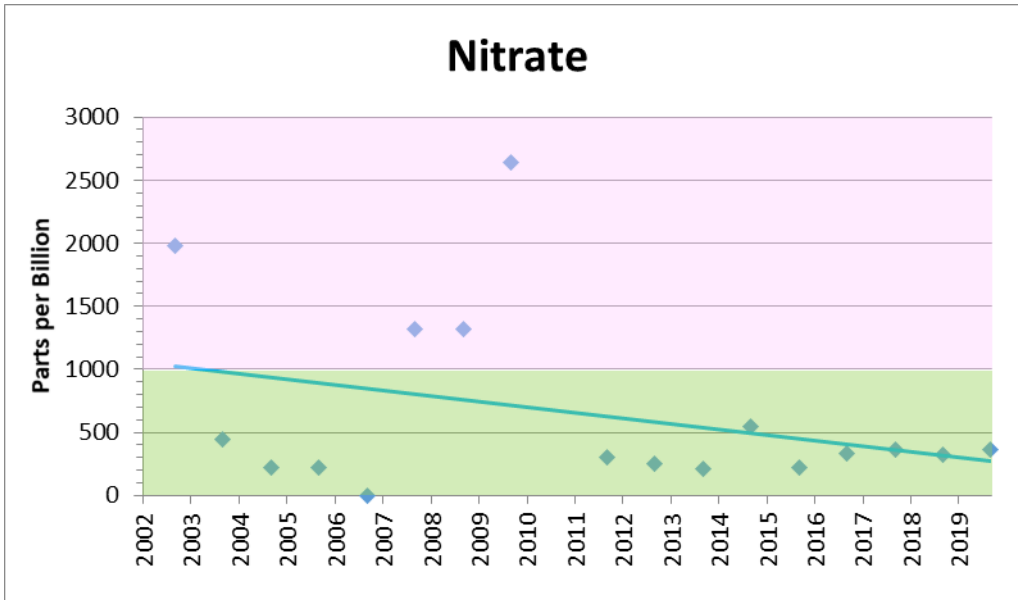


Phosphate (ppb)					
	April	May	June	July	August
Site 1	60	40	50	60	60
Site 2	30	50	50	50	50
Site 3	40	60	60	60	50
Site 4	70	50	50	50	40
Season Average					52

**Target Range: 0 – 100 ppb**

Phosphate is the form of phosphorus that is usable to aquatic plants. The phosphate data also showed a slightly upward trend since we began testing this parameter in 2011. Despite the increase, the phosphate remained within the target range and helped to limit plant and algae growth in the lake.

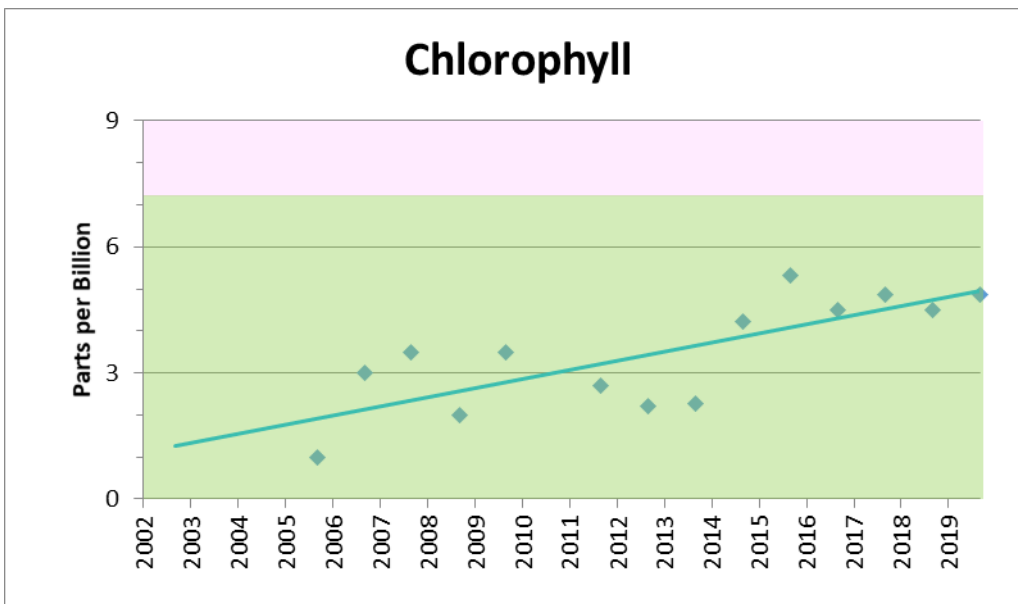




Nitrate (ppb)					
	April	May	June	July	August
Site 1	308	308	352	352	396
Site 2	264	352	352	308	484
Site 3	352	440	352	352	352
Site 4	396	440	308	308	396
Season Average					359

**Target Range: 0 – 1,000 ppb**

Nitrate is another vital nutrient for the growth of aquatic plants. Water Quality Investigators obtained high results early in the testing history, so the recent data resulted in a downward historical trend. It is important residents continue to be conscience of their property and practices to ensure more nutrients do not enter the lake.



Chlorophyll (ppb)					
	April	May	June	July	August
Site 1	2.7	3.4	4.0	7.7	8.9
Site 2	1.5	3.5	3.9	8.0	8.5
Site 3	2.3	2.9	3.4	7.2	8.6
Site 4	1.5	2.3	4.3	6.9	9.0
Season Average					5.0

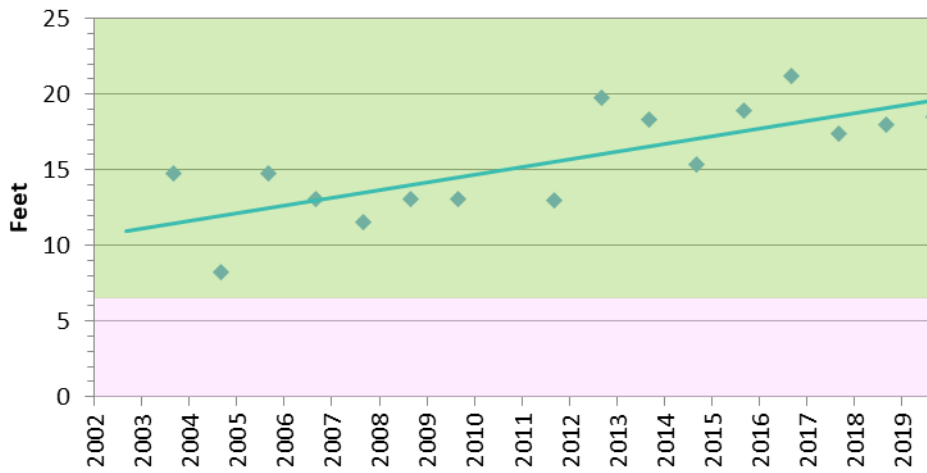
**Target Range: 0 – 7.2 ppb**

Chlorophyll concentrations were tested as an indicator of plant production, primarily algae in the water column. Over the testing history, the results increased steadily, which is expected with invasive plants in the lake, such as Eurasian Milfoil. This reinforces the urgency for responsible land management in the watershed to prevent additional phosphorus from entering the lake.





## Transparency

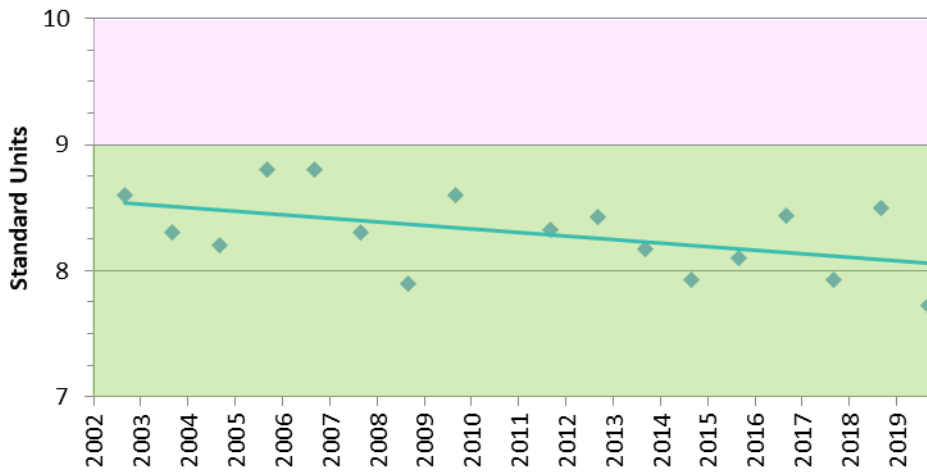


Transparency (feet)					
	April	May	June	July	August
Site 1	23.8	15.1	18.4	16.1	17.8
Site 2	26.0	15.3	16.3	12.7	19.2
Site 3	27.3	21.5	16.8	13.3	19.5
Site 4	21.7	15.9	21.3	13.5	18.7
Season Average					18.5

**Target Range: > 6.5 feet**

Transparency was affected by different factors including total dissolved solids, total suspended solids, algae growth, and rain frequency and amount. Overall, the transparency of the lake increased over the testing history.

## pH



pH (Standard Units)					
	April	May	June	July	August
Site 1	8.0	8.2	7.5	7.6	7.9
Site 2	8.0	7.8	7.5	7.6	8.0
Site 3	7.9	7.5	7.4	7.6	8.0
Site 4	7.9	7.2	7.2	7.7	7.9
Season Average					7.7

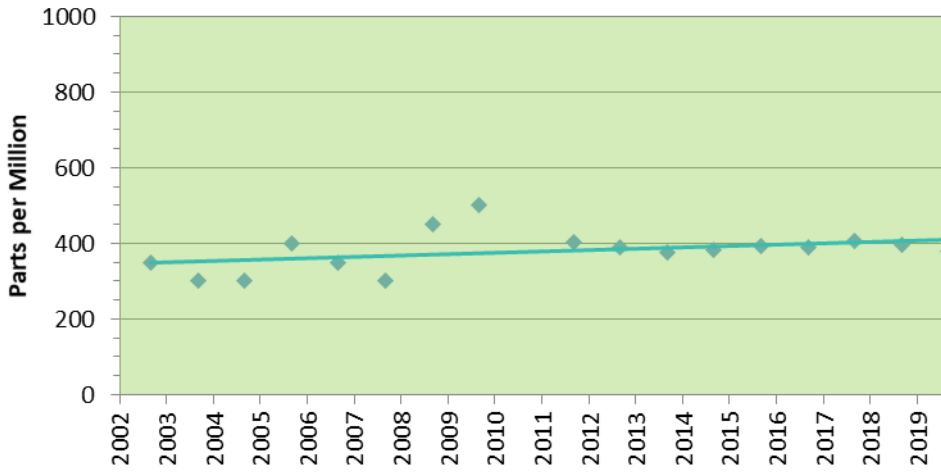
**Target Range: 7.0 – 9.0 S.U.**

There was a slight decrease in pH over the testing history, but it stayed in the target range of 7 to 9. We will look for the pH to remain level in future years. If the pH ever drops drastically, we will look for the cause of that change in order to mitigate the trend.





## Total Dissolved Solids

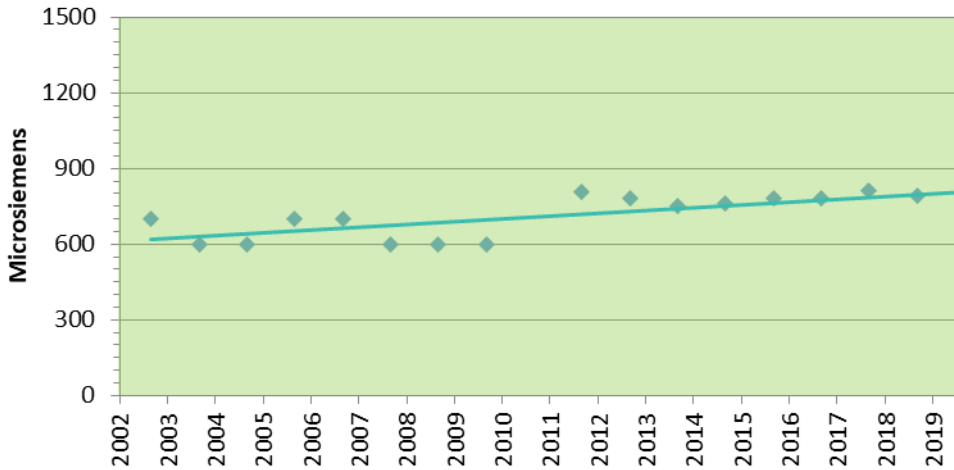


Total Dissolved Solids (ppm)					
	April	May	June	July	August
Site 1	401	384	368	359	366
Site 2	401	388	371	368	369
Site 3	400	387	373	366	368
Site 4	399	387	373	365	368
Season Average					378

**Target Range: 0 – 1,000 ppm**

The Total Dissolved Solids show a very slight upward trend over the testing history. This is a common trend for inland lakes as they accumulate more substances from their watershed and the atmosphere.

## Conductivity



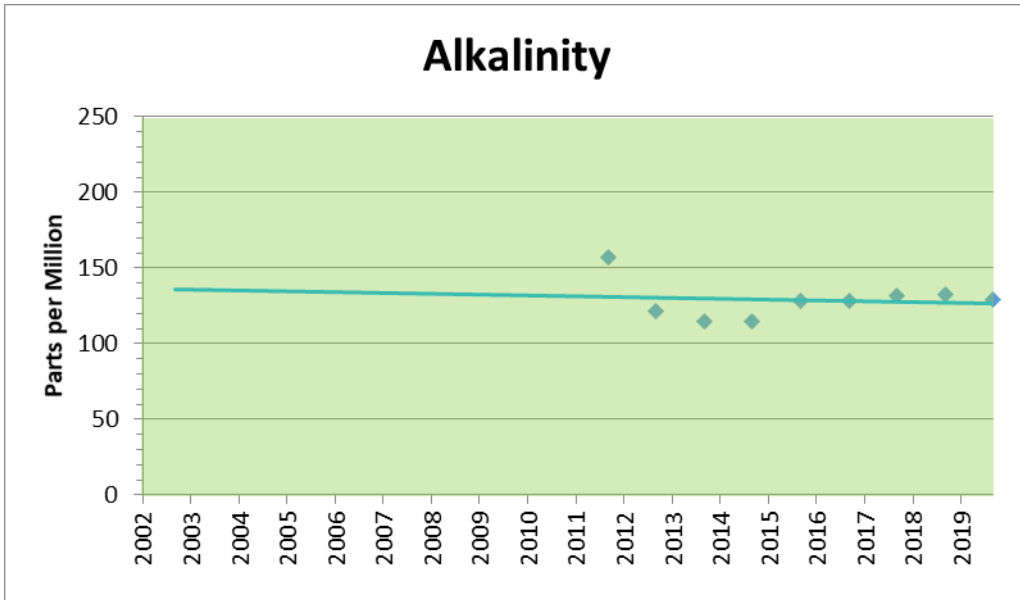
Conductivity (µS)					
	April	May	June	July	August
Site 1	802	769	736	717	732
Site 2	802	775	742	736	739
Site 3	800	775	747	732	738
Site 4	799	775	746	731	736
Season Average					756

**Target Range: 0 – 1,500 µS**

Like the TDS, Conductivity showed a very slight upward trend. Conductivity is an extension of TDS and measures the number of ionic molecules in the water (which conduct electricity, usually salts).



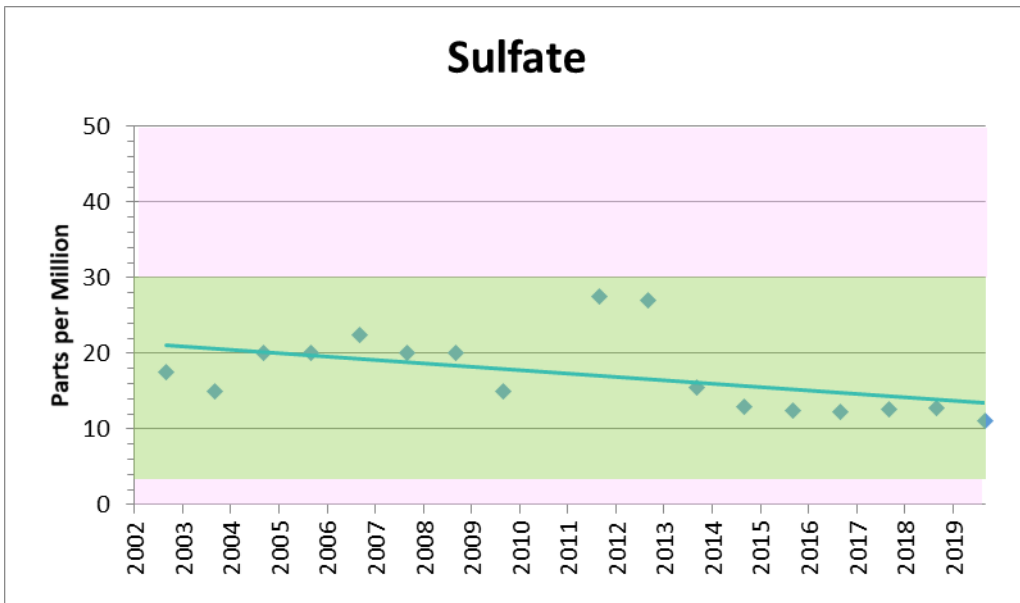




Alkalinity (ppm)					
	April	May	June	July	August
Site 1	124	139	139	123	120
Site 2	133	137	136	123	123
Site 3	127	132	137	122	122
Site 4	124	132	136	130	129
Season Average					129

**Target Range: 0 – 250 ppm**

The alkalinity decreased slightly since we started testing this parameter in 2011. Alkalinity works as a buffer to stabilize the pH when foreign substances enter the lake.



Sulfate (ppm)					
	April	May	June	July	August
Site 1	11.0	11.0	11.0	12.0	12.0
Site 2	11.0	10.0	11.0	11.0	12.0
Site 3	11.0	10.0	11.0	11.0	12.0
Site 4	11.0	10.0	11.0	11.0	12.0
Season Average					11.1

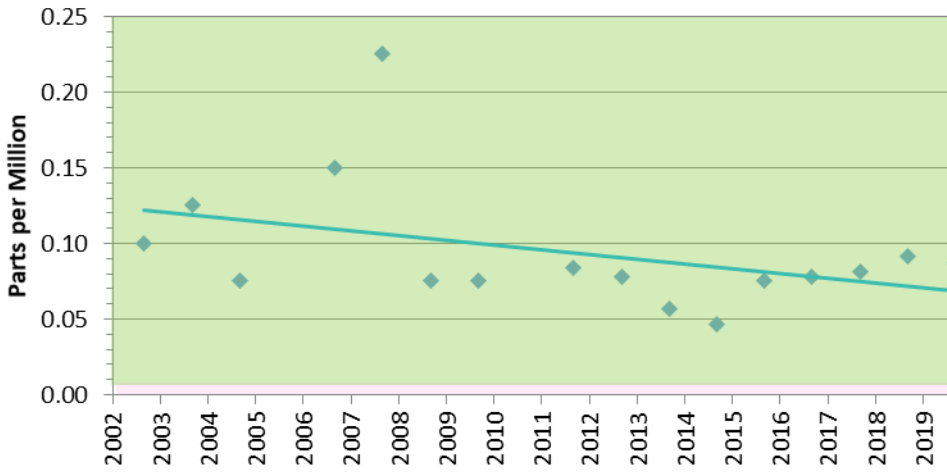
**Target Range: 3 – 30 ppm**

Calcium sulfate and magnesium sulfate are common minerals in surface water, so some sulfate should be present. Elevated levels of sulfate can indicate pollution. Over the testing history, sulfate remained within the target range and decreased toward the bottom of the target range.





## Fluoride

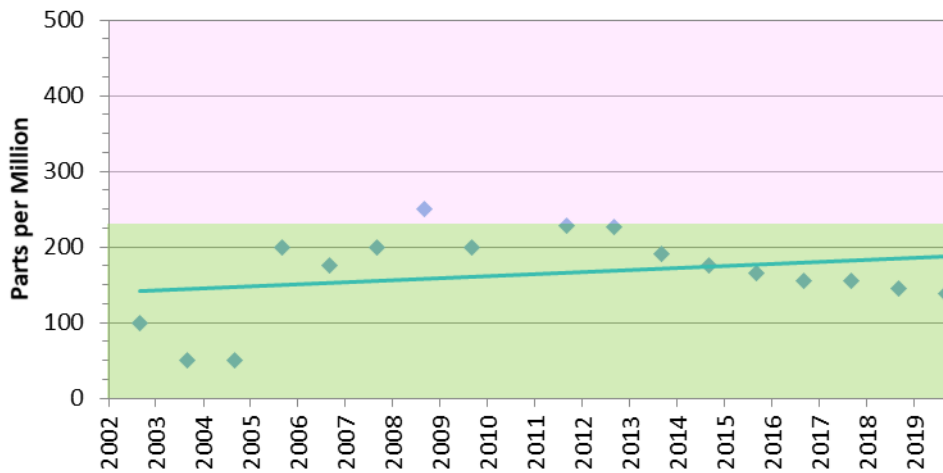


Fluoride (ppm)					
	April	May	June	July	August
Site 1	0.09	0.08	0.07	0.10	0.08
Site 2	0.07	0.07	0.11	0.10	0.07
Site 3	0.07	0.11	0.09	0.08	0.08
Site 4	0.07	0.11	0.11	0.09	0.09
Season Average					0.09

### Target Range: 0.01 – 0.30 ppm

Fluoride occurs naturally in ground water, so some may be present in the lake surface water. Elevated levels can indicate pollution, but are not physiologically harmful. Over the testing history, fluoride remained within the target range and decreased toward the bottom of the target range.

## Chloride



Chloride (ppm)					
	April	May	June	July	August
Site 1	153	146	132	129	129
Site 2	162	149	131	133	121
Site 3	156	151	135	127	120
Site 4	160	143	132	125	120
Season Average					138

### Target Range: 0 – 230 ppm

Chloride is a major anion found in water. This substance may be due to the natural process of water passing through salt formations in the earth or may be evidence of the intrusion of pollution from industrial processes or road salting. Over the testing history, the chloride concentrations increased slightly, despite a downward trend in recent years. Despite the upward trend, the concentrations remained within or close to the target range.





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## Conclusion

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Overall, the water quality of Lower Long Lake was excellent again this year. Only one parameter, Total Phosphorus, was outside the target range. All other parameters remained within their target ranges and some were the best we saw across the state. Our only concerns at this time were the historical increases of phosphorus and chlorophyll. It is vital that everyone within the watershed take steps to limit their nutrient input to the lake.

Despite a heavily developed watershed and homes surrounding the lake, Lower Long Lake is an exemplary water resource with great water quality. There will always be areas that the quality of the water could improve, but the lake remains among the best that we test. You should take pride in this lake and continue your hard work in improving it.

Thanks for choosing LakePro,

Peter Filpansick  
Director of Lake Management  
Aquatic Biologist





### Analysis Information

Temperature:	The water temperature directly affects the amount of oxygen that can dissolve into the water. The temperature of surface waters is not indicative of the entire water column.
Transparency:	The ability of light to penetrate the water column is determined by the amount of dissolved and suspended particles in the water. Although aesthetically desirable, transparent water allows increased light to reach the lake bed and may result in vegetation growth.
pH:	pH is a measure of acidity or alkalinity. pH is a general measure of lake health and can roughly indicate the range of other measurements such as alkalinity and hardness.
TDS:	Total Dissolved Solids is the amount of all organic and inorganic substances in the water in a molecular or ionized state. Higher values generally indicate richer and more productive water. Lower values usually indicate cleaner and less productive water.
Conductivity:	Conductivity is a measure of the ability of water to conduct electricity. Dissolved ions in the water increase conductivity, thus TDS and Conductivity are closely related.
Alkalinity:	Alkalinity refers to the ability of the water to neutralize acids, mainly through the hydrogenation of carbonate ions. Therefore the alkalinity is expressed as "ppm as $\text{CaCO}_3$ ". However, other basic molecules in the water can also contribute to alkalinity.
Dissolved Oxygen:	D.O. is a measure of the amount of oxygen dissolved in the water. This oxygen is available to fish and other animals for respiration. Vegetation generally increases DO, particularly during the day and early evening. Animals and other respiring organisms consume the oxygen, mostly during the day. Oxygen is also added to the lake through wave action, rain, fountains and aerators.
Total Phosphorus:	Phosphorus is an essential nutrient for plant growth. However, concentrations exceeding 100 ppb can impair the water and results in nuisance vegetation growth.
Phosphates:	Phosphate is the form of phosphorous that is most readily available to plants and algae.
Nitrate:	Nitrogen is also essential for plant growth. Nitrate is the predominant form of nitrogen in water. Excessive nitrate concentrations may also result in pollution and increased vegetation.
Chlorophyll-a:	Chlorophyll-a is a direct measurement of the amount of green pigment produced by plants and phytoplankton. This indicates the amount of plant growth and is used to calculate a Trophic State Index.
Sulfate:	Sulfate occurs naturally as minerals, such as calcium sulfate and magnesium sulfate. In fresh water, sulfate is usually the second or third most abundant anion. Other sources of sulfate include water material from pulp mills, steel mills, food processing operations, and municipal wastes. Under low oxygen conditions, sulfate can be reduced to hydrogen sulfide gas, which smells like rotten eggs.
Fluoride:	Fluoride may occur naturally or be added to public drinking water supplies.
Chloride:	Chloride is one of the major anions found in water and sewage. The presence of chlorides may be due to water passing through salt formations in the earth or pollution from industrial processes, domestic wastes, or road salt. The salt content of water affects the





distribution of plant and animal life in an aquatic system, based on the amount of slat they can tolerate.

**Fecal Coliforms:** Non-fecal coliforms are naturally found as soil organisms. Fecal Coliforms, such as *E. coli*, are coliforms found in the intestines of warm-blooded animals and humans. The presence of fecal coliforms indicates contamination from either animals or humans.

### **Trophic States**

**Oligotrophic:** Water is very clear. Nutrient levels are generally low. Plant and algae productivity are also low. Sufficient dissolved oxygen in the bottom, cooler waters allows cold-water fish to survive, such as salmon and trout.

**Mesotrophic:** Water is moderately clear. Nutrient levels are slightly elevated. Plant and algae productivity are present, but generally not a nuisance. Oxygen and temperature in the lower portion of the lake allow walleye and perch to survive.

**Eutrophic:** Water is not clear due to high nutrients levels, increased turbidity, and excessive algal growth. There is no oxygen in the bottom, cooler waters, restricting the lake to warm water species, such as bass and bluegill.

**Hypereutrophic:** Nutrient levels are extremely high, promoting very high algae productivity. Blue-green algae blooms are likely. High turbidity and algae growth make the water opaque. Little plant growth is restricted to invasive plants. The only fish that can survive this environment are rough fish, such as carp, catfish, and mudminnows.

**Sample Sites:**

